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High pressure discharge lamp

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The invention relates to a high-pressure discharge lamp, which is in particular suited for use in plant growing irradiation.

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The absorption of green leaves is strongest in the blue and the red part of the spectrum. Photons (quanta) between 400 and 700 nm are determining the rate of 5 photosynthesis. As absorption of these photons is the driving force for photosynthesis, a spectral quantum yield of photosynthesis has been derived by MCree (The action spectrum, absorptance and quantum yield of photosynthesis in crop plants, Agric. Meteorol. 1971/1972, 9. 191 - 216) and refined by Sager et al. (Light Energy Utilization Efficiency for Photosynthesis, Transactions of the ASAE, General Edition, 1982, 25/6, 1737 - 1746). These studies learn that the yield of photosynthesis is high over a large region having relative 10 maxima in the blue and the red part of the spectrum. High intensity discharge lamps with Na or NaI efficiently emit radiation in particular in the region of the NaD-line at 589 nm, where the absorption of the chlorophyll is strong. Particularly, high-pressure sodium (called SON or alternatively HPS) lamps are therefore presently used for assimilation lighting in green houses. SON lamps reach luminous efficacies between 100 and 150 lm/W and photon flux sensitivity up to 1.9 μ mol/(Ws). High intensity discharge lamps with comparable luminous efficacies are lamps on the basis of NaI and CeI₃ fillings. In EP 0 896 733 a metal-halide system with NaI and CeI3 is described, which can reach efficacies between 130 and 174 lm/W. The luminous efficacy decreases when Li is added. In US 6,147,453 a lamp with NaI, CeI3, and LiI filling is described, which only reaches luminous efficacies between 100 and 135 lm/W.

For an efficient support of plant growth, lamps must generate light very efficiently in the region where the photosynthetic yield is maximal.

The main drawback of the known lamp with the filling comprising the combination of NaI, CeI₃, and Lil is that it emits a considerable amount of light in the green region of the spectrum, where the photosynthetic yield is lowest. Although it has a high luminous efficacy, it is less suitable for stimulation of plant growth than lamps on the basis of Na or NaI, which emit more efficiently in the red part of the spectrum. Both the lamp with a

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filling comprising NaI/CeI3 and the one comprising Na, Ce and Li halides have the drawback of being susceptible to demixing phenomena of the filling during lamp operation.

The main disadvantage of SON lamps is that they emit mainly around 589 nm although the plants still absorb photons very efficiently up to approximately 700 nm. The conversion of electrical power into photons of SON lamps is therefore not optimal with respect to the plant absorption spectrum.

In literature a lamp for promoting plant growth is proposed having a ceramic discharge vessel comprising Hg, LiI in an amount of between 0.02 to 4.2 mg/cm3, and an excess of Li to compensate for effects of corrosion. Due to the relative large amount of Hg the spectrum of the lamp has a relative large quantity of the emitted light in the green part of the spectrum. This is a drawback as it is not really effective in plant growth.

According to the invention the high-pressure discharge lamp has a discharge vessel, which besides a buffer gas comprises an excess amount of substantially LiI as metal halide or of a mixture of LiI and NaI, the lamp having a coldest spot temperature Tes during normal operation of at least 1200K. Normal operation of the lamp is understood in this respect stable operation at a lowest power and on a corresponding voltage for which the lamp is designed. As buffer gas Hg is frequently used. Besides the discharge vessel may comprise a rare gas like Ar, Kr and Xe or a mixture of thereof, which promotes starting and can also have a buffer gas capacity. In particular Xe also has a buffer gas capacity with increased fill pressures. The discharge vessel can be made of ceramic or made of quartz or quartz glass. Ceramic is meant in this respect to be translucent mono or densely sintered poly crystalline metal oxide, like Al_2O_3 , Y_2O_3 , $Y_3Al_5O_{12}$ (YAG) and densely sintered metal nitride, like AlN. A 150 W Lil-lamp according to the invention with mercury as buffer gas and a ceramic alumina discharge vessel, for instance, emits 15 - 20 % of its radiation in the blue region between 400 and 500 nm and about 75 % in the red region between 600 and 700 nm, which 25 are surprisingly high percentages. The emission of the lamp thus matches the absorption spectrum of green plants surprisingly very well, which match is much better than that of a high-pressure sodium lamp, where only up to 10 % is emitted in the blue region and at most about 40 % in the red region. The high percentage of blue light was in itself unexpected because the main lines of Li are at 611 and 671 nm, respectively. A further surprising 30 advantage of the lamp according to the invention is that no traces of serious corrosion are recorded.

A coldest spot temperature T_{es} below 1200K results in partial vapour pressures of Li becoming so low that the percentage in the blue region strongly reduces and

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consequently the radiation contribution of the Hg gets more important. The latter however results in an increasing part of the radiation being emitted in the green part of the spectrum, which is however ineffective with regard to plant growth.

Whereas the use of LiI as a filling component generally means a reduction of 5 the luminous efficacy (see above), it surprisingly turns out that the energy conversion of a lamp according to the invention is at least comparable or even better than that of a comparable known lamp. For a 150W lamp with a filling comprising Na or NaI the energy conversion efficacy is about 27 %, which value increases to almost 30 % for the invented 150 W lamp described above. This increase is surprising and unexpected. Despite the higher blue fraction of the Li spectrum, the photon flux per input power (in $\mu mol/(W^*s))$ of the invented lamp turned out to be even 10% higher than is the case with the comparable lamp having a filling of Na or NaI.

The new lamp therefore provides a higher energy and higher photon efficiency as well as a spectrum, which is better adapted to the plant absorption and photosynthetic quantum yield.

The above and further aspects of the invention will be explained in more detail below with reference to a drawing, in which:

Fig. 1 schematically shows a lamp according to the invention,

Fig. 2 shows in detail the discharge vessel of the lamp in accordance with

Fig. 1, and

Fig. 3 shows a spectrum of a lamp according to the invention compared with a non-invented lamp.

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Fig 1 shows a discharge lamp according to the invention having a ceramic wall. Fig. 1 shows a metal halide lamp provided with a discharge vessel 3 having a ceramic wall, which encloses a discharge space 11 containing an ionizable filling. Two electrodes whose tips are at a mutual distance EA are arranged in the discharge space, and the discharge vessel has an internal diameter Di at least over the distance EA. The discharge vessel is closed at one side by means of a ceramic projecting plug 34, 35 which encloses a current lead-through conductor (Fig. 2: 40, 41, 50, 51) to an electrode 4, 5 positioned in the discharge vessel with a narrow intervening space and is connected to this conductor in a gastight

manner by means of a melting-ceramic joint (Fig. 2: 10) at an end remote from the discharge space. The discharge vessel is surrounded by an outer bulb 1, which is provided with a lamp cap 2 at one end. A discharge will extend between the electrodes 4, 5 when the lamp is operating. The electrode 4 is connected to a first electrical contact forming part of the lamp cap 2 via a current conductor 8. The electrode 5 is connected to a second electrical contact forming part of the lamp cap 2 via a current conductor 9. The discharge vessel, shown in more detail in Fig. 2 (not true to scale), has a ceramic wall and is formed from a cylindrical part with an internal diameter Di which is bounded at either end by a respective end wall portion 32a, 32b, each end wall portion 32a, 32b forming an end surface 33a, 33b of the discharge space. The end wall portions each have an opening in which a ceramic projecting plug 34, 35 is fastened in a gastight manner in the end wall portion 32a, 32b by means of a sintered joint S. The ceramic projecting plugs 34, 35 each narrowly enclose a current leadthrough conductor 40, 41, 50, 51 of a relevant electrode 4, 5 having a tip 4b, 5b. The current lead-through conductor is connected to the ceramic projecting plug 34, 35 in a gastight manner by means of a melting-ceramic joint 10 at the side remote from the discharge space. The electrode tips 4b, 5b are arranged at a mutual distance EA. The current lead-through conductors each comprise a halide-resistant portion 41, 51, for example in the form of a Mo-Al₂O₃ cermet and a portion 40, 50 which is fastened to a respective end plug 34, 35 in a gastight manner by means of the melting-ceramic joint 10. The melting-ceramic joint extends over some distance, for example approximately 1 mm, over the Mo cermet 40, 41. It is 20 possible for the parts 41, 51 to be formed in an alternative manner instead of from a Mo- $\mathrm{Al_2O_3}$ cermet. Other possible constructions are known, for example, from EP-0 587 238. A particularly suitable construction was found to be a halide-resistant coil applied around a pin of the same material. Mo is very suitable for use as a highly halide-resistant material. The parts 40, 50 are made from a metal whose coefficient of expansion corresponds very well to 25 that of the end plugs. Nb, for example, is for this purpose a highly suitable material. The parts 40, 50 are connected to the current conductors 8, 9 in a manner not shown in any detail. The lead-through construction described renders it possible to operate the lamp in any burning nosition as desired.

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Each of the electrodes 4, 5 comprises an electrode rod 4a, 5a which is provided with a coiling 4c, 5c near the tip 4b, 5b. The projecting ceramic plugs are fastened in the end wall portions 32a and 32b in a gastight manner by means of a sintered joint S.

In a practical embodiment the ceramic wall is made from alumina and the thus formed discharge vessel has a diameter of 4 mm and a length of 36 mm. In fig 3 the spectrum 10

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of a lamp, of which the metal halide filling substantially comprises an excess amount of 8mg Lil is shown with curve 1. Of a comparable non-invented lamp, which filling comprises NaI instead of Lil the spectrum is shown with curve 2. In both lamps the filling of the discharge vessel also comprises 3.8mg Hg as buffer gas and 300mbar Ar/Kr. The lamp according to the invention has a coldest spot temperature $T_{\rm es}$ during normal operation of 1376K. The coldest stop temperature $T_{\rm es}$ was directly measured by means of an infrared camera. The spectrum of the non-invented lamp is equivalent to the spectrum of an ordinary HSP lamp. From the shown spectra it is clear that the blue fraction in the spectrum 1 of the Lil comprising lamp is much higher than in the HPS spectrum 2. It is also clearly shown that the spectrum 1 emits much more radiation in the region from 600 to 700nm than the spectrum 2. A further advantage of the lamp according to the invention is that its visible lumens are more than a factor 2 lower than in case of a HPS or of a NaI comprising lamp of comparable wattage. Thus lighting for plant growing, so called assimilation lighting, results in less illumination of the surroundines.

The invented lamp described above is used in assimilation lighting in green houses. In an experiment the effect on tomato plants, chrysanthemum plants and potted roses is investigated. In a first area plants were illuminated with lamps according to the invention having a total flux input per unit area of 118 µmol/s. As comparison in a second area plants were illuminated with an HSP lamp having a total flux output of 122 µmol/s. After an illumination period of only 18 hours there was an equal or even a slightly better growing of the plants in the first area, in particular with respect to the tomato plants. As the photon efficiency of a lamp comprising the filling of LiI is about 15% higher than the photon efficiency of a comparable HPS lamp, the assimilation lighting through the invented lamp requires less nominal lamp power for achieving equal growing results.

Comparison of some lamp characteristics of a lamp according to the invention with lamps not according to the invention is given below.

The lamp according to the invention has a filling of 3.8mg Hg and LiI. The lamp has a nominal power of 150W. The photon flux per unit power of this lamp is 1.5 μ mol/(W*s).

For a comparable lamp in which the halide is NaI the phton flux per unit power is 1.35 μ mol/(W*s).

A HPS lamp with a nominal power of 150W has a photon flux per unit power of 1.29 μ mol/(W*s).

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 $\label{lem:condition} Copies of the lamp according the invention have been operated for 5000 hours \ .$ without showing any serious corrosion.